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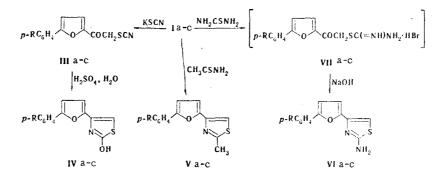
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(5-Aryl-2-furyl)glyoxals and 4-(5-aryl-2-furyl)thiazoles were synthesized from 5-aryl-2-bromoacetylfurans.

Continuing our research on arylfurans, we have studied the possibility of using the 5-aryl-2-bromoacetylfurans (Ia-c) that we previously [1] synthesized for the synthesis of (5-aryl-2-furyl)glyoxals. (5-Aryl-2-furyl)glyoxal hydrates (IIa-c) were obtained in 60-90% yields by the action of dimethyl sulfoxide (DMSO) on I.

 $p \cdot RC_6H_4 \xrightarrow{(CH_3)_2SO} p \cdot RC_6H_4 \xrightarrow{(COCHO \cdot H_2O)} p \cdot RC_6H_4 \xrightarrow{(COCHO \cdot H_2O)} COCHO \cdot H_2O$ $I a - C \qquad II a - C$ $a R = Br; b R = CI; C R = NO_2$

We also studied the possibility of using I in the synthesis of thiazole systems containing a 5-arylfuryl group in the 4 position of the thiazole ring. The reaction of alcohol solutions of bromo ketones (Ia-c) with potassium thiocyanate, thioacetamide, or thiourea gave, respectively, 2-hydroxy- (IVa-c), 2-methyl-(Va-c), or 2-aminothiazoles (VIa-c). 2-Amino-4-[5-(p-nitrophenyl)-2-furyl]thiazole (VIc) was obtained in acetone with preliminary isolation of isothiuronium hydrobromide VIIc.



EXPERIMENTAL

[5-(p-Bromophenyl)-2-furyl]glyoxal Hydrate (IIa). A solution of 1.7g (3 mmole) of Ia in 7 ml of DMSO was heated at 50 deg for 4 h and allowed to stand overnight. The next day, 50 ml of water was added to the mixture, and the precipitated IIa was removed by filtration. Glyoxals IIb,c were similarly obtained (Table 1).

5-(p-Bromophenyl)-2-(thiocyanatoacetyl)furan (IIIa). A solution of 3.53 g (10 mmole) of Ia in 40 ml of alcohol was added to a solution of 1.54 g (14 mmole) of potassium thiocyanate in 5 ml of alcohol, and the mixture was heated for 1 h with stirring on a boiling-water bath, after which the precipitate was removed by

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| TABLE 1. | (9-A) | TABLE 1. (5-Aryl-2-furyl)glyoxals (IIa | l)glyc | oxals (IIa-c) | | | | | | | | | | | | | | |
|-------------------|--------|--|------------------|---|--------------|-----------------------|--------------|-----------|-----|--------|------------------|------|----------|-----------------------------------|-------------------------------------|------------------------|--------------|--------|
| | | | | | | Fou | Found % | | | | Calc., % | ., % | | IR spectrum | um | Is AN | UV spectrumb | |
| Com- pound | а | mp, °C | R ₁ a | Empirical formula | | н <mark>о</mark> Н | hal- ogen | z | Ś | | n hal- ogen | | 2 Z | $v_{c=0} \cdot cm^{-1}$ | v ₀₁₁ , cm ⁻¹ | λ _{m ax} , nm | ह ह | Yield, |
| IIa Br | Br | 131-135 | | | ****** | 3,0 | 26,5 | | | | | 26,9 | | 1655 m 1680 | 3380-3420 330; 226 4,47; 4,13 | 330; 226 | 4,47; 4,13 | 65 |
| Thiosemicarb (| Dazone | 225—226 126—127.5 ^c | | C ₁₃ H ₁₀ BrN ₃ O ₂ S C ₁₂ H ₇ ClO ₃ · H ₂ O | 44,2 57,1 | 3,8 | | 11,8 | 9,0 | 57,1 3 | 2,8 3,6 14 | | 1,9 9,1 | | 3380-3418 330; 228 4,41; 4,14 | 330; 228 | 4,41; 4,14 | 69 |
| Thiosemicarbazone | azone | 229 - 230 | 0,66, | C13H10CIN3O2S | | _ | | | | 6.0 | 3,3 | _ | 10,4 | | 0010 | | | c 1 |
| Dihvdrate | So. | 84868 146 148 d | | C ₁₂ H ₇ NO ₅ · H ₂ O | | 0 1 1 1 | | 2,2 | | 1 9 1 | 4, O | | ν. Ο | 1000-1080 V.Dr 3280-3400 352; 222 | 3280-3400 | 352; 222 | 4,58; 4,22 | 28 |
| Thiosemicarbazone | azone | 230-231 | | | | 31.5 | | 17,8 10,3 | | 0,01 | 2 | | 7,6 10,1 | | | | | |
| | - | | | - , | - | | - | - | _ | . ' | - | | - | | - | - | - | |

^aSilufol, development in UV light. ^bIn ethanol. ^c With decomposition. ^dFrom water. ^eEther. ^fEther-ethyl acetate (5:1). ^gFrom alcohol. ^hEther-petroleum ether (4:1).

TABLE 2

| | Yield, | 06 | 60 | 75 | 20 | 68 | 65 | 06 | 65 | 83 |
|--------------------------|-----------------------|--|--|--|---------------------------------------|---------------------------------------|---|--|----------------------|--|
| UV spectrum ^a | lg e | 4,46; 4,19 | | | | | 4.37: 4.08 | | | |
| nV sp | λ _{max} , nm | 333; 261 | | | 328; 233 | 324; 230 | 386; 272 | 335; 272 | 334; 272 | 256; 300; 406 |
| | s | 6,9 | 11,5 | 1,11 | 10,0 | 11,6 | 11,2 | 10,01 | 11,6 | 11,2 |
| | z | 4,3 | | 9,7 | 4,4 | | 9,8 | 24,9 | 10,1 | 14,6 |
| Calc., % | halogen | 24,8 | 12,8 | | 24,9 | 12,9 | | 8,7 | 12,8 | |
| 0 | Ŧ | . 2,5 | 2,9 | 2,8 | 3,1 | 3,6 | 3,5 | 2,8 | 3,3 | 3,2 |
| | D | 48,5 | 56,2 | 54,2 | 52,5 | 61,0 | 58,7 | 48,6 | 56,4 | 54,3 |
| | s | 9,8 | 11,4 | 11,1 | 9,8 | 11,7 | 10,8 | 9'6 | 11,8 | 11,1 |
| % | z | 4,6 | | 9,8 | 4,4 | | 10,1 | 8,6 | 6'6 | 14,6 |
| Found, % | hal- ogen | 24,4 | 12,6 | | 25,2 | 12,9 | | 24,5 | 12,9 | |
| | Ħ | 2,6 | 2,8 | 2.8 | 3,2 | 3.5 | 3.5 | 2,8 | 3,3 | 3,1 |
| | υ | 48,1 | 55,8 | 53,8 | 52,1 | 60.7 | 59.0 | 48,6 | 56.2 | 54,3 |
| | Empirical formula | C ₁₃ H ₈ BrNO ₂ S | C ₁₈ H ₈ CINO ₈ S | C ₁₃ H ₈ N ₂ O ₄ S | C ₁₄ H ₁₀ BrNOS | C ₁ ,H ₁₀ CINOS | C ₁ ,H ₁₀ N ₂ O ₃ S | C ₁₃ H ₉ BrN ₉ OS | ClaHaCIN2OS | C ₁₃ H ₉ N ₃ O ₅ S |
| | mp, °C | 248—249 ^b | $249-252^{D}$ | $266-269^{b}$ | 139-140 ^c | 135.5-137 ^c | 174.5-177 ^c | 192.5-193.5 ° | 172-173 ^C | $228 - 230^{\circ}$ |
| | R, | HO | HO | HO | CH, | CH, | CH, | NH, | NFI. | NH2 |
| | ĸ | Br | 0 | , on | Br | 10 | őZ | Br | Ū | NO2 |
| | - moo | IVa | IVb | Nc | Va | ζЪ. | Vc | VIa | VIb | VIC |

^aIn ethanol. ^bFrom acetic acid. ^cFrom alcohol.

filtration to give 3.2 g (98%) of IIIa with mp 141-142 deg (from alcohol). Found: N 4.6; S 9.6%. $C_{13}H_8Br-NO_2S$. Calculated: N 4.3; S 9.9%.

 $\frac{5-(p-Nitrophenyl)-2-(thiocyanatoacetyl)furan (IIIc).}{and had mp 151.5-153°} (from alcohol). Found: C 53.8; H2.7; N 9.8; S 10.8\%. C₁₃H₈N₂O₄S. Calculated: C 54.2; H 2.8; N 9.7; S 11.1\%.$

2-Hydroxy-4-[5-(p-bromophenyl)-2-furyl]thiazole (IVa). A mixture of 2.42 g (7 mmole of IIIa in 40 ml of glacial acetic acid, 1 ml of water, and 0.3 ml of concentrated sulfuric acid was heated with stirring for 30 min on a boiling-water bath, after which it was cooled, and IVa was removed by filtration. Hydroxy-thiazoles IVb,c (Table 2) were similarly obtained.

2-Amino-4-[5-(p-bromophenyl)-2-furyl]thiazole (VIa). A mixture of 1 g (13 mmole) of thiourea in 3 ml of water and 4.34 g(13 mmole) of Ia in 50 ml of alcohol was heated for 2 h with stirring on a boilingwater bath, after which it was cooled and treated with 2.5 g of sodium hydroxide. The mixture was then poured into 100 ml of water, and the precipitated VIa was removed by filtration. 2-Aminothiazole VIb (Table 2) was similarly obtained.

<u>S-{2-[5-(p-Nitrophenyl)-2-furyl]-1-oxoethyl}</u> isothiuronium Hydrobromide (VIIc). A solution of 0.25 g (4 mmole) of thiourea in 15 ml of acetone was added to a solution of 1 g (3 mmole) of Ic in 4 ml of acetone, and the mixture was refluxed for 30 min. It was then cooled and filtered to give 1.1 g (89%) of hydrobromide VIIc, which had mp 308-310° (dec.) after washing with acetone. Found: C 40.3; H 2.9; Br 20.5; S 8.1%. $C_{13}H_{11}N_{3}O_{4}S \cdot HBr$. Calculated: C 40.4; H 3.1; Br 20.7; S 8.3%.

2-Amino-4-[5-(p-nitrophenyl)-2-furyl]thiazole (VIc). Ammonium hydroxide (20%) was added to a suspension of 1 g (2 mmole) of hydrobromide VIIc in 5 ml of water until the mixture was alkaline, and the precipitated VIc was removed by filtration (Table 2).

LITERATURE CITED

1. A. F. Oleinik, T. I. Vozyakova, G. A. Modnikova, and K. Yu. Novitskii, Khim. Geterotsikl. Soedin., 1448 (1972).